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Abstract. Lincoln Laboratory of Massachusetts Institute of Technology has developed a technique known as dynamic photoacoustic spectroscopy (DPAS) that could enable remote detection of trace gases via a field-portable laser-based system. A fielded DPAS system has the potential to enable rapid, early warning of airborne chemical threats. DPAS is a new form of photoacoustic spectroscopy that relies on a laser beam swept at the speed of sound to amplify an otherwise weak photoacoustic signal. We experimentally determine the sensitivity of this technique using trace quantities of SF<sub>6</sub> gas. A clutter-limited sensitivity of  $\sim$ 100 ppt is estimated for an integration path of 0.43 m. Additionally, detection at ranges over 5 m using two different detection modalities is demonstrated; a parabolic microphone and a laser vibrometer. Its utility in detecting ammonia emanating from solid samples in an ambient environment is also demonstrated. © The Authors. Published by SPIE under a Creative Commons Attribution 3.0 Unported License. Distribution or reproduction of this work in whole or in part requires full attribution of the original publication, including its DOI. [DOI: 10.1117/1.OE.53.2.021103]

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#### **1** Introduction

Chemical threats (in a variety of forms) are an area of pressing national security concern.<sup>1</sup> Airborne threats of interest are gaseous or aerosol based. Additionally, certain solid hazardous materials emanate vapor signatures that potentially could be used to detect the solid. A detection system that is able to rapidly detect and/or identify a chemical threat is of great interest as it allows appropriate countermeasures to be taken. An ideal technique detects the threat rapidly and with high sensitivity. A wide-area coverage rate is also generally preferred. Numerous optical techniques have been investigated to perform these functions. Generally, an optical spectroscopy approach is taken whereby key absorption features are used to identify threats of interest. Differential optical absorption spectroscopy (DOAS) identifies absorption features in the ultraviolet and visible regions using a broadband optical source and a dispersive receiver (usually grating based). It has been used to detect  $NO_3^-$  at concentration path lengths of 2 ppb-meter; however, to do so a remote retroreflector was required.<sup>2</sup> A similar technique known as differential infrared absorption lidar (DIAL) identifies infrared absorption features. It does not necessarily require a monostatic configuration, but there is a cost in sensitivity. It has been used to detect 2 ppm-meter  $SF_6$  plumes without the use of a remote reflector.<sup>3</sup> Lidar systems measuring Raman shifts have also been used, but with significantly less sensitivity than the other techniques.<sup>4</sup> The focus of many techniques to date has been on gases; however, aerosols may also be identifiable via their absorption features.<sup>5</sup>

Dynamic photoacoustic spectroscopy (DPAS), in a manner similar to DIAL, identifies target materials based on characteristic absorption features. Unlike DIAL or DOAS, however, the detected signal is acoustic, not optical. Since the acoustic background is very different than the optical background, the background-limited sensitivities will be different and may potentially favor an acoustic technique. DPAS is a variant of photoacoustic spectroscopy (PAS),<sup>6</sup> which also detects photoacoustic signals and has demonstrated very high detection sensitivities. PAS has achieved a sensitivity of 50 ppt using an acoustically resonant cell.<sup>7</sup> It relies on the conversion of an optical absorption into a propagating acoustic wave via the heating of the local air. The optical to acoustic coupling is generally weak, however; thus amplification is required for good signal to noise. This is generally achieved using closed resonant cells, not easily amenable to the standoff detection problem. Enhanced gain, and thus improved PAS detection range, has been achieved by increasing the receiver size (e.g., via parabolic microphones).

DPAS achieves amplification of the photoacoustic signal without the need for a resonant cavity by sweeping the laser beam at the speed of sound, thus generating a shock-wavelike response. The concept was recently demonstrated using SF<sub>6</sub> as a test gas.<sup>8</sup> These proof-of-concept studies characterized the acoustic waveform, which resembled that of a shock wave when the beam was swept at Mach 1. The dependence of the signal amplitude on source velocity fits a simple moving monopole model well. A related experiment<sup>9</sup> utilized a sonically swept beam to generate pressure waveforms near an air/water interface. The theoretical analysis of this phenomenon was only partially successful at describing their results, implying the need for further analysis. In this work we quantify the sensitivity of the DPAS technique and discuss the factors limiting the sensitivity, i.e., receiver noise and ambient acoustic clutter. We extend the earlier work (which utilized a remote laser and local microphone) to a standoff configuration in which both laser and receiver are distant from the detected gas. Finally, we demonstrate the utility of the technique to detect the gas emanating from solid materials.

#### 2 Detection Sensitivity

The experimental system has been described in detail elsewhere.<sup>8</sup> For our detection sensitivity experiments, we used a  $CO_2$  laser to interrogate a target gas (SF<sub>6</sub>) contained within a loosely sealed anechoic chamber. The chamber walls were coated with anechoic foam to minimize reflections and simulate an open air environment. The gas concentrations were determined using an Online Technologies (now MKS Instruments, Andover, Massachusetts) Fourier transform infrared (FTIR) spectrometer that sampled the chamber gas. The laser was tuned to an absorption feature of interest  $(10.591 \ \mu m)$  of the target gas  $(SF_6)$  within the longwave infrared. The laser beam was rotated by a spinning mirror, and two omnidirectional microphone detectors (Earthworks M30; bandwidth  $\sim 50$  kHz; sensitivity = 0.32 V/Pa when combined with our preamplifier) were placed at opposite sides on the far end of the chamber such that the laser beam rotated toward (approaching) one and away from (receding) the other.

Initially, the laser was chopped at 3.1 kHz, and the response of both microphones was measured for different laser sweep speeds. As expected based on the simple Doppler-shift physics of a moving source term, the measured signals are very different depending on whether the beam approaches or recedes from the given microphone. The measured spectral response of both microphones is shown in Fig. 1 for several different beam speeds. The dashed lines of the figure are the expected frequencies based on the Doppler shift of a source moving with the speed of the beam at the target range. They fit the measured responses quite well. As expected based on the physics of a moving source,<sup>10</sup> the magnitude of the signal at the receding microphone decreases as the beam speed increases, becoming nearly negligible at Mach 1. In contrast, the signal magnitude at the approaching microphone increases as the beam speed approaches Mach 1. It also broadens spectrally. At Mach 1, the signal magnitude peaks; spectrally it spans roughly 30 kHz with a peak near 10 kHz. The corresponding

temporal waveform at Mach 1 is a single pulse with width of  $\sim 0.1$  ms.

Our measurements of detection sensitivity used the approaching microphone at the optimum beam speed of Mach 1 (mirror rotation rate of 20 Hz). The laser beam integrates over a gas path length of 0.43 m prior to reaching the microphone. Previous measurements indicate that the acoustic waveform is several inches in spatial extent. In order to capture all this energy we increase the collection area of our microphone by centering it within a 9-in diameter parabolic collector. We used a 3.5 W cw CO<sub>2</sub> beam since previous measurements show the signal is maximized with no chopper. Internal fans were used to ensure homogenization of the gas. Signals were averaged over a 60-s data acquisition period (1200 scans through the sample). Note that the laser beam was inside the sample chamber only  $\sim 3\%$  of the time it was rotating; so the active interrogation time was 1.8 s. The peak-to-peak microphone signals are displayed as a function of gas concentration (as measured via FTIR) in Fig. 2. The detection sensitivity of the FTIR was only a few ppb, at which point the DPAS signal-to-noise ratio (SNR) was  $\sim 100$ . In order to estimate the DPAS system sensitivity, we recorded the ambient signal with no gas present. Since the experiment took place in a laboratory, a fume hood was the largest source of acoustic noise. The clutter level is shown in Fig. 2. Linearly extrapolating the  $SF_6$  data to the clutter level yields a detection limit of  $\sim 100$  ppt. Given our 0.43 m path length and the absorption of  $SF_6$ (0.5 cm<sup>-1</sup> Torr<sup>-1</sup>), this corresponds to a minimum measured absorbance of  $1.7 \times 10^{-6}$ . An absorbance of  $8.8 \times 10^{-9}$  has been reported<sup>11</sup> for PAS using resonant cells (5 s integration); however, these are inherently not standoff measurements. The receiver noise levels of the Earthworks microphone and a lower-noise microphone (Bruel & Kjaer, Denmark) are also displayed in Fig. 2, indicating that the detection is clutter limited.

Previous measurements<sup>8</sup> indicate that the strength of the DPAS response scales linearly with gas absorbance and



Fig. 1 Power spectral density of the DPAS signal at various beam speeds for both the approaching and receding microphones. The dashed lines are predicted Doppler shifts. Inset is a closeup of the low-frequency data.



Fig. 2 DPAS signal versus SF<sub>6</sub> gas concentration. Blue line is a linear extrapolation of the SF<sub>6</sub> data. Estimated NH<sub>3</sub> signal versus concentration (black line). Horizontal lines indicate clutter and receiver noise levels.

integration path length; thus we can extrapolate the DPAS response to other materials using known absorptivities. Using the absorptivity<sup>12</sup> of NH<sub>3</sub> at 10.33  $\mu$ m (0.03 cm<sup>-1</sup> Torr<sup>-1</sup>), we estimate the DPAS response to NH<sub>3</sub> over a 1 m path (see Fig. 2). Extrapolating to the clutter level indicates an NH<sub>3</sub> sensitivity of ~1 ppb. In Sec. 4, we test these conclusions via direct measurements of NH<sub>3</sub>.

#### **3 Remote Measurements**

The performance of the DPAS technique in an ambient environment at ranges exceeding 5 m was investigated using two different sensors: a parabolic microphone and a laser vibrometer. A laser vibrometer<sup>13</sup> could be quite useful when coupled with the DPAS concept because it detects the acoustic signal near the source itself, rather than relying on a back-propagating acoustic signal as would be the case when using a remote parabolic microphone. Laser vibrometers function as interferometers, sending out a (typically) eyesafe beam of light and examining the phase shifts of the backscattered return. Pressure disturbances along the path of the beam induce changes in the local index of refraction, which are detectable as phase shifts by the vibrometer. We used a Polytec OFV 5000 with VD06 decoder yielding a bandwidth of 20 kHz and a sensitivity of 1 mm/s/V, where 1 mm/s is the gas particle speed. Assuming standard atmospheric conditions, this yields a sensitivity of 2.5 V/Pa. For these measurements, we removed the top from our sample chamber and aimed the vibrometer above the chamber, scattering it off a surface at the far end of the chamber (Fig. 3). The vibrometer beam was aimed 6 in. above the  $CO_2$  beam. Note that the vibrometer beam is thus in a direction off-axis from the acoustic wave propagation direction. We released a quick (~seconds) burst of SF<sub>6</sub> gas, with a concentration we estimate (based on the local microphone response) to be in the single-digit ppm. The responses of both the vibrometer and the local microphone (no parabolic collector) are displayed in Fig. 4. Also displayed are their responses in the absence of SF<sub>6</sub> gas. The local microphone records a sharp peak 1.7 ms after the beam enters the chamber, consistent with a signal (and laser beam) moving at the speed of sound toward the local receiver. The vibrometer records a somewhat broader signal occurring 0.25 ms later than the microphone signal. The delay is consistent with an acoustic signal propagating upward several inches. The fact that the vibrometer signal is broader than that of the microphone is likely the result of the fact that the vibrometer responds to acoustic signals along its entire path length,





**Fig. 3** Experimental setup of remote vibrometer measurement. Both the vibrometer and  $CO_2$  laser are at a distance from the gas being detected. The vibrometer beam is directed 6 in. above the  $CO_2$  beam and aligned above the close-range microphone.



Fig. 4 Response of (a) laser vibrometer at 6 m standoff (upper; red) and (b) local microphone (lower; blue) to a release of  $SF_6$  gas. Black curves are responses with no gas present.

whereas the microphone responds to acoustic disturbances only at its range.

In a setup similar to that of the vibrometer, a parabolic collector (52 cm diameter;  $\sim$ 30 dB gain at 20 kHz) with an Earthworks M30 microphone was placed 5.1 m from the far wall. The parabolic microphone (unlike the vibrometer) relies on the detection of a back-propagating signal to the detector, which results in a signal that is delayed relative to the local microphone (Fig. 5). The observed delay of 14 ms is consistent with speed of sound travel to the remote parabolic microphone.

As noted in the Introduction, traditional PAS requires direct sampling of a gas within a closed resonant cell and is not amenable to standoff detection. A direct comparison of our DPAS results to PAS performance is thus not possible. We can, however, use our data collected using a stationary, chopped beam (see Fig. 1) as a surrogate for a PAS response. Note that our surrogate PAS response does not have the



Fig. 5 Response of (a) remote parabolic microphone at 5.1 m standoff (upper; red) and (b) local microphone (lower; blue) to a release of SF<sub>6</sub> gas.

Table 1	Comparison of	of DPAS to	open cell PA	S at a	range of 6 m
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	DPAS (parabolic mic)	DPAS (vibrometer)	PAS
SF <sub>6</sub> concentration	~1 ppm	~7 ppm	15 ppm
Integration time	20 scans	100 scans	100 scans
Range	5.1 m	6 m	~1 cm
SNR	29	17	200
Normalized SNR (6 m; 10 ppm; 100 scans)	$4.7 \times 10^{2}$	$2.4 \times 10^{1}$	$3.7 \times 10^{-4}$

advantage of a closed resonant cell. Table 1 compares the DPAS response at a distance [both via vibrometer (Fig. 4) and parabolic microphone (Fig. 5)] to the surrogate PAS response. The different measurement conditions are listed. For the DPAS data, the SNR was calculated as the mean signal over the standard deviation of the background. For the PAS data, the SNR was calculated as the ratio of the power spectral densities at 3.1 kHz. The final row displays SNR normalized to 6 m (assuming an inverse-squared range dependence), 10 ppm, and 100 scan averages; it indicates many orders of magnitude advantage for the DPAS technique. While the PAS technique could be improved via the use of a parabolic collector, the collection area would have to be exceedingly large to increase the SNR to a level comparable to the demonstrated DPAS performance.

#### 4 Detection of Ammonia in an Ambient Environment

We used a solid source of ammonia vapor to demonstrate the utility of the DPAS technique to detect important vapors in an open, ambient environment. The solid-state source relied on an acid-base reaction using an ammonium salt to produce NH<sub>3</sub>. Solid samples were placed in open trays extending over a 4 in.  $\times$  36 in. line (Fig. 6). A local microphone with 9 in. collector (same as Sec. 2) was placed at the end of the array. The CO<sub>2</sub> laser was tuned to an NH<sub>3</sub> absorption line at 10.33  $\mu$ m and scanned 0.5 in. above the sample at Mach



Fig. 6 Test setup used to detect  $NH_3$  vapors emanating from a solid source of ammonia in an ambient environment.



Fig. 7 Microphone response to solid sources of NH<sub>3</sub> in an ambient environment.

1 (12 Hz rotation). An Online Technologies FTIR spectrometer sampled the local head space, recording varying concentrations of NH<sub>3</sub> as high as 300 ppb. The microphone response averaged over 12 scans (1 s data acquisition; 0.042 s on target gas) is displayed in Fig. 7. The  $\sim$ 20 mV peak at 3.5 ms is due to the NH<sub>3</sub> absorption (verified via background measurements for which no peak was observed). The SNR is estimated at ~20 based on the ~1 mV background signal (no material present). Our limiting sensitivity for NH<sub>3</sub> detection based on these ambient measurements (SNR = 1) is thus ~15 ppb. We can compare this ambient result to our estimates of Fig. 2 by assuming SNR scales as the square root of integration time. If the ambient data were scaled to the 1.8 s integration ( $43 \times$  longer) of Fig. 2, then the detection sensitivity is  $\sim 2$  ppb, which is in fairly reasonable agreement with the 1 ppb estimate of Fig. 2.

#### 5 Conclusions

Laboratory measurements indicate that the DPAS technique may be useful for the standoff detection of trace vapors. Our anechoic chamber measurements indicate clutter-limited sensitivities of ~100 ppt (SF<sub>6</sub>; 0.43 m integration) and  $\sim$ 1 ppb (NH<sub>3</sub>; 1 m integration). Additionally, we demonstrate in an ambient laboratory environment that DPAS can be useful at standoff ranges of at least several meters. Of the two sensing modalities tested, the vibrometer is more amenable to monostatic standoff detection. In order to ascertain the technique's field utility, outdoor measurements are planned in which realistic clutter levels will be assessed.

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#### References

- 1. "Chemical, and biological defense," Lincoln Lab. J. 17(1), 1-198 (2007).
- U. Platt and J. Stutz, *Differential Optical Absorption Spectroscopy Principles and Applications*, Springer-Verlag, Berlin (2008).
   T. Fujii and T. Fukuchi, *Laser Remote Sensing*, Taylor & Francis, Boca
- Raton, Florida (2005).
- C. Philbrick et al., "Remote sensing of chemical species in the atmosphere," presented at Proc. of Fourth Symp. on Lidar Atmospheric Applications, Phoenix, Arizona (11-15 January 2009).
- 5. K. Gurton et al., "In situ aerosol spectroscopy for a variety of nerve agent stimulants using flow-through photoacoustics," Appl. Opt. 46(25), 6323-6329 (2007).
- A. C. Tam, "Applications of photoacoustic sensing techniques," *Rev. Mod. Phys.* 58(2), 381–431 (1986); C. K. N. Patel and A. C. Tam, "Pulsed photoacoustic spectroscopy of condensed matter," *Rev. Mod. Phys.* **53**(3), 517–550 (1981).
- 7. M. A. Gondal, "Laser photoacoustic spectrometer for remote monitor-
- M. A. Obladi, Last photosite spectrometric for reinformed in formation of atmospheric pollutants," *Appl. Opt.* **36**(15), 3195–3201 (1997).
   C. M. Wynn et al., "Dynamic photoacoustic spectroscopy for trace gas detection," *Appl. Phys. Lett.* **101**(18), 184103 (2012).
   Y. H. Berthelot and I. J. Busch-Vishniac, "Thermoacoustic radiation of the photoacoustic application of the photoacoustic a
- sound by a moving laser source," J. Acoust. Soc. Am. 81(2), 317-327 (1987)
- 10. P. M. Morse and K. U. Ingard, Theoretical Acoustics, p. 724, Princeton University Press, Princeton (1986).
- 11. M. B. Pushkarsky, M. E. Webber, and C. K. N. Patel, "Ultra-sensitive ambient ammonia detection using CO<sub>2</sub>-laser-based photoacoustic spectroscopy," *Appl. Phys. B* 77(4), 381–385 (2003).
  12. R. P. Patty et al., "CO<sub>2</sub> laser absorption coefficients for determining
- ambient levels of O<sub>3</sub>, NH<sub>3</sub>, and C<sub>2</sub>H<sub>4</sub>," Appl. Opt. 13(12),
- 2850–2854 (1974). 13. L. A. Jiang et al., "Laser vibrometry from a moving ground vehicle," *Appl. Opt.* **50**(15), 2263–2273 (2011).

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